

Studies on the Total Synthesis of Hainanolide (VII) - Analysis of the Relative Stereochemistry of key Intermediate 2

Yan Wu LI, Liang HUANG*

Institute of Materia Medica, Chinese Academy of Medical Sciences & Peking Union Medical College, Beijing 100050

Abstract: The relative stereochemistry of an intermediate **2** in the total synthesis of hainanolide **1** was studied by various NMR spectroscopic techniques.

Keywords: Hainanolide, NOE, ^1H , ^{13}C -NMR, HMBC, HMQC, spectra.

The natural product, hainanolide¹ **1**, also under the name harringtonolide² demonstrated antitumor and antiviral activities in preliminary test. Its structure was determined by X-ray diffraction. The total synthesis of **1** was reported recently by Mander³. A different scheme of its synthesis has been studied in our laboratory⁴. Here the determination of the stereochemical structure of the key intermediate **2** in the synthesis was reported.

HMBC and HMQC spectra identified the skeleton and H, C correlation of compound **2** as shown in **Figure 2**. The results of ^1H and ^{13}C spectra were listed in **Table 1**. NOE revealed that A, B, C three rings being *cis* fused as that in hainanolide and with 5a-COOCH₃ and C₃-OH at the same side as H_{2a} and C₄-OCH₃ at back side being at the right positions for building lactone ring and oxobridge.

Based on the data of coupling constant (J Hz: $\text{H}_{2a}/\text{H}_3=12$ $\text{H}_3/\text{H}_4=7.2$, $\text{H}_4/\text{H}_5=0$) shown in **Table 1**, the dihedral angles of C_{2a}-H/C₃-H, C₃-H/C₄-H and C₄-H/C₅-H were evaluated about 180°, less than 180° and 90° respectively. The molecular model was thus constructed which seems to be comfortable as deformed boat for ring B, chair for A and envelop for C (**Figure 3**) to reduce possibly steric effect between substituents. It coincides nicely with what obtained from calculation using Alchemy III. Further confirmation of the stereochemical structure by X-ray analysis will be carried out later.

Figure 1

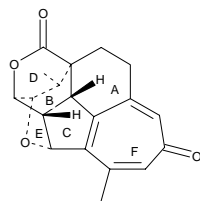


Figure 2

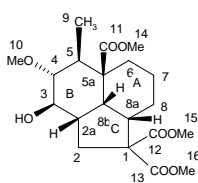


Figure 3



Table 1 ^1H (300MHz), and ^{13}C (75MHz) NMR data for **2**(CDCl_3)

No	δ_{C}	δ_{H}	No	δ_{C}	δ_{H}
1	64.34		8b	37.42	3.13(dd, 1H, $J_{8b,2a}=7$, $J_{8b,8a}=12$)
2	35.98	2.38(dd, 1H $_{\alpha}$, $J_{2\alpha,2a}=9$, $J_{2\alpha,2\beta}=14.5$)	9	19.40	0.91(d, 3H, $J_{9,5}=7.5$)
		2.60(dd, 1H $_{\beta}$, $J_{2\beta,2a}=8.5$, $J_{2\beta,2\alpha}=14.5$)			
2a	37.42	2.10(dddd, 1H, $J_{2a,2\beta}=8.5$, $J_{2a,2\alpha}=9$, $J_{2a,3}=12$, $J_{2a,8b}=12$)	10	57.38	3.42(s, 3H)
3	78.39	3.90(dd, 1H, $J_{3,4}=7$, $J_{3,2a}=12$)	11	170.23	
4	90.92	2.94(d, 1H, $J_{3,4}=7$)	12	171.97	
5	41.50	2.27(q, 1H, $J_{5,9}=7.5$)	13	176.81	
5a	49.66		14	52.42	3.70(s, 3H)
6	32.85	1.98(m, 1H $_{\alpha}$)1.57-1.65(m, 1H $_{\beta}$)	15	52.04	3.72 (s, 3H)
7	21.63	1.57-1.65(m, 1H $_{\beta}$) 1.04(m, 1H $_{\alpha}$)	16	52.76	3.75 (s, 3H)
8	25.24	1.44(m, 1H $_{\beta}$) 0.90(m, 1H $_{\alpha}$)	OH		1.57-1.65(m, 1H)
8a	43.48	2.94(dd, 1H, $J_{8a,8b}=7$, $J_{8a,8\beta}=7$, $J_{8a,8\alpha}=12$)			

Table 2 NOESY 1D of compound **2** (500MHz in CDCl_3)

Irradiation H	Correlation peaks
H ₂	H _{2a} , H ₃
H _{2a}	H ₂ , H ₄ +H _{8a} , C ₅ -Me, H _{8b}
H ₃	H ₂ , H _{6\alpha}
H ₄ +H _{8a}	H _{2a} , H _{8b} , H _{8\beta} , C ₅ -Me, C ₄ -OMe
H ₅	H _{6\beta} , C ₅ -Me, C ₄ -OMe
H _{8b}	H _{2a} , H ₄ +H _{8a} , C ₅ -Me

References

1. Buta, *et al.*, *J. Org. Chem.*, **1978**,43,1002.
2. N. Sun, *et al.*, *Acta. Pharm. Sinica*, **1979**,14,39.
3. L. N. Mander, *et al.*, *J. Am. Chem. Soc.*, **1998**, 120, 1914.
4. Luyan Zhang, *et al.*, *Chin. Chem. Lett.*, **1996**, 7, 892.

Received 8 January, 2002